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# Determination of The Physical, Physio-Chemical and Chemical Properties of Gilaburu Fruits (Viburnum opulus) Dried by Convectional Drying Technique<sup>#</sup>

Fatma Fulya Dal<sup>1,a,\*</sup> Erkan Karacabey<sup>1,b</sup>

 $^{I}$ Food Engineering Department, Engineering Faculty, Süleyman Demirel University, 32200 Isparta, Turkiye \*Corresponding author

#### ARTICLE INFO

#### ABSTRACT

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Research Article

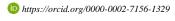
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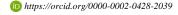
Our country has many species in fruit and vegetable cultivation. One of them is Gilaburu fruit (Viburnum Opulus), which has spread to different regions, especially in Kayseri. Longer preservation of products in fruits and vegetables; Drying process has been applied since ancient times to prevent deterioration caused by microbial and/or biochemical changes. One of them and the most common one is convectional drying in other words drying in a drying chamber under controlled hot air stream. In this study, Gilaburu fruits were obtained from a local producer in Kayseri. Drying processes were carried out in a convectional oven at 60, 70 and 80°C in three repetitions. The effects of different drying conditions on the drying

characteristics of Gilaburu fruits were examined. Drying time for Gilaburu fruits decreased with increasing temperature. As a color parameter total color change with respective to fresh fruit ( $\Delta E$ ) was investigated. The results indicated that it remained almost identical with temperature change. Similar trend was observed for rehydration rate. Titration acidity and pH values showed compatible changes. With temperature increase, titration acidity increased, as pH decreased. As textural properties, skinhardness and elasticity of dried fruits were measured. As hardness increased, elasticity decreased with temperature elevation. Total phenolic content and antioxidant activity of dried samples were also examined. The results showed that there were increases in both of them with an increase in temperature.











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# Introduction

Gilaburu fruit having valuable nutrients is processed to different food products such as fruit juice, nectar, vinegar, jam (Taşkın et al., 2019).

For longer shelf life of fruit and vegetable products, drying process has been used since ancient times due to its preservation effect against deterioration caused by microbial and/or biochemical changes (Alçay et al., 2015). Sun drying is one of the drying methods used by manufacturers. Besides of advantages like being low cost and practical application, this method leads to nutrient and quality losses in dried products (Midilli, 2001). Thus, producers focus on alternative ones. One of them and the most common one is the convectional drying in other words drying in a drying chamber under controlled hot air stream (Taşova et al., 2019). Convectional drying system consists of an oven equipped with a heater, a shielded impeller fan to distribute the hot air stream over single or more trays (Oliveira et al., 2016).

In this study, drying of Gilaburu fruits was aimed in a conventional oven at different temperature levels. Drying characteristics, the physical, physico-chemical and chemical properties of Gilaburu fruits were also investigated.

# **Materials and Methods**

#### Materials

Gilaburu fruits used in this study were obtained from Kayseri. Fruits were harvested from plants in the same garden. Then, they were moved to laboratory of Food Engineering Department of Süleyman Demirel University and kept in a refrigerator until drying process.

# **Drying Method**

Before drying process, Gilaburu fruits were washed and sorted to achieve uniform and homogeneous samples. For convectional drying processes, Mikrotest brand MKD-250 model tray-type drying oven equipped with a temperaturecontrolled was used. Drying processes were carried out at temperature levels of 60, 70 and 80°C. The mean air flow rate was 1.3 m/s in drying chamber. Gilaburu fruits were speared as a single layer on a tray. Amount of sample used for each drying trial was constant and determined according to the system's capacity. During drying processes, moisture content of drying samples was monitored and process was finalized when sample's moisture content reduced below 15% (Maisnam et al., 2017). Afterwards, dried gilaburu fruits were packed in a vacuum packaging material in a vacuum device (Vacuum Packaging Machine, DZ-350/MS, CHINA) and stored at -18°C until analysis.

#### Color Measurement

The color values of the dried samples were measured using a colorimeter (Model no: NH310, Portable Colorimeter, CHINA) and expressed as CIE ( $L^*$ ,  $a^*$ ,  $b^*$ ) color parameters (Keskin et al., 2017). Color measurement for each drying trial was carried on 10 dried samples and each was mean of the 3 locations of each one. Surface color change with drying process was evaluated compared to fresh Gilaburu fruit as a total color change ( $\Delta E$ ). Results were given as mean  $\pm$  standard error of all measurements for each drying trial.

# Rehydration Rate

A fixed amount (4 g) of dried sample was immersed in water at room temperature. For certain period, rehydrated samples were taken out and excess surface water was removed and samples were weighted and re-immersed in water (Khraisheh et al., 2004). Analysis was terminated when the weight of samples remained constant. The rehydration rate was calculated as the ratio of the final weight of the sample to its initial weight. The result for each drying trial was in the mean of three replicates.

### **Titration Acidity**

For titration acidity, a fixed amount (4 g) of sample was homogenized in a fixed volume (100 ml) of distilled water. It was then filtered through coarse filter paper. Filtrate was titrated with 0.1 N NaOH and titration acidity of each sample was expressed in % citric acid (Cemeroğlu, 2019).

# Solid/Liquid Extraction

The solid/liquid ratio for the extraction process was 1/10 (g/ml). The extraction solvent was pure methanol. Fruit samples and solvent were homogenized with Ultra-Turrax (IKA T18 basic, Germany) for 1 minute in a beaker. The homogenized solution was stirred on a magnetic stirrer at room temperature for 2 hours. After extraction, the liquid phase was separated from the solid sample phase by passing through filter paper, and then stored at -18 °C until analysis (Fang et al., 2002).

# Total Amount of Phenolic Compounds of Extracts

The total amount of phenolic compounds of the extract was determined by using Folin-Ciocalteu spectrophotometric method developed by Singleton and Rossi (1965).

Briefly, 0.04 ml of phenolic extract was mixed with 2.4 ml of deionized (DI) water and then 0.2 ml of Folin-Ciocalteu reagent was added. After 5 minutes, 0.6 ml of saturated Na<sub>2</sub>CO<sub>3</sub> solution was added. The resulting mixture was left to stand for 2 hours for color development and absorbance at 765 nm was recorded using a spectrophotometer (T70+UV/VIS, PG Instruments, UK). Gallic acid was used as a standard to generate the calibration curve for quantification. Results were calculated as mg gallic acid equivalent per g dried matter.

# Antioxidant Activity of Extracts

Antioxidant activity analysis was performed according to 2,2-diphenyl-1-picryl-hydrazil (DPPH) free radical

scavenging activity assay (Dorman et al., 2003). Antioxidant activity of extracts from fresh and dried Gilaburu samples was measured as % inhibition. Result was expressed as "mg trolox equivalent per g dry matter".

#### Texture Analysis

Skin hardness (g. force) and elasticity of dried products and fresh Gilaburu fruit were determined using a texture analyzer (TA-XT2 Texture Analyzer, Stable Microsystems, Godalming, UK) according to puncture test method. A 2 mm needle prob was used and prob speed was set as 1 mm / sec (Barragán-Iglesias et al., 2019). Result was the mean of the corresponding textural values of three samples for each trial.

# Statistical Analysis

Significant differences (P<0.05) between the means in the analysis data were evaluated with one-way ANOVA. Data were expressed as mean  $\pm$  standard error (Nunes et al., 2015). Statistical analysis was performed in Minitab version 17.1.0.

#### **Results and Discussion**

# Drying of Gilaburu Fruit by Convectional Drying

Drying times were recorded for three different drying temperature values (60, 70 and 80 °C) during convectional drying of Gilaburu fruit samples (Table 1). Drying time for Gilaburu fruits decreased with increasing temperature. Literature studies have also indicated the increase in drying rate and decrease in drying time with the increase in drying temperature (Kavak et al., 2006; Lee and Kim, 2009; Rafiee et al., 2010).

Table 1. Drying temperature and the corresponding drying time values

Drying	Time
60°C	2663 min
70°C	856 min
80°C	420 min.

The favorable effect of temperature increase is related to heat and mass transfer phenomena taking place during drying. As temperature increased, temperature difference driving heat transfer increased and more energy was transferred to drying sample per unit time. As a result of more energy, the amount of water evaporated from the solid structure per unit time increased and drying was accelerated (Zhu and Shen, 2014). In addition, temperature dependent change in relative humidity of the drying air also affected mass transfer and thus drying rate (Masud et al., 2019). Increasing temperature decreased relative humidity of drying air, thus, water transfer from fruit materials to the air stream was accelerated.

# Color Value

As a color parameter, total color change with respective to fresh fruit ( $\Delta E$ ) was examined. The calculated total color change values for corresponding drying trials were shown in Table 2. The results indicated that there was color change with drying, but irrespective of temperature value. Color change in Gilaburu fruits with drying was thought to

be due to an adverse thermal effect on color pigments and also resulting enzymatic and none-enzymatic reactions (Skrede et al., 2000).

In a study in which hot air drying of food materials was examined at different temperatures, it was reported that color losses at low air velocity were caused by prolonged drying time (Tontul et al., 2019). Zielinska et al. (2016) have studied the drying of blueberries and reported similar results to present ones. In that study, it was observed that convectional drying produced dehydrated samples with great overall change in color values depending on temperature level and the most significant change in color attributes was determined for drying trial at 60°C.

# Rehydration Rates

Rehydration rates of dried food materials are widely used as a quality index (Qing-guo et al., 2007). Rehydration results were given in Table 2. Drying temperature did not create any effect on rehydration ratio of dried samples. In other words, it can be concluded that there is no significant difference between total color changes of dried samples depending on processes temperature in the studied range (P>0.05).

#### **Texture**

Skin-hardness values of fresh and dried Gilaburus at different temperature levels were measured and results were given in Table 3. It was observed that the skinhardness values of dried products increased with an increase in drying temperature. It was associated with case hardening on fruit surface layer which gets stronger with temperature increase as a result of accelerated moisture transfer. Similarly, Zielinska et al. (2016) showed that the hardness of dried blueberries increased with convective drying at elevated temperature levels, and reported the highest hardness value for drying at 80°C. This temperature dependent change was explained with structural changes in fruit matrix. It was said that the highest values of hardness convectional drying at 80°C could be attributed to the presence of a thick surface layer. Furthermore, they have discussed that higher water vapor pressure inside the sample at elevated temperature levels

resulted in an accumulation of flesh under the skin which is the main reason for higher hardness values. Elasticity values of fresh and dry Gilaburus at different temperature levels of conventional drying were measured and given in Table 3. The results indicated that an increase in drying temperature caused a decreasing elasticity of fruit samples. This was thought to be related to structural changes which occur during drying process and get stronger with elevating temperature as a result of increasing crispness depending on moisture removal.

# Titration Acidity and pH

In this study titration acidity (%) and pH values of fresh and dried Gilaburu samples were examined to determine the effect of conventional drying process carried out at different temperatures. The measured results were shown in Table 3. It was determined that the titration acidity values increased with drying process due to the proportional change in organic acid content as a result of moisture removal. It can be seen that there is no any strong variation in titration acidity (%) of dried samples depending on temperature level. However, compared to that value for samples dried at 80°C, it was found lower for dried samples obtained at 60°C. Khiari et al. (2019) also reported that a decrease in titratable acidity of dried raisin was favored by low air dehydration temperature and long process time. Being coincidence with titration acidity results, pH values for corresponding drying trials displayed similar variation trends compared to fresh one and there was no any significant difference in between each other.

# Total Amount of Phenolic Compounds

Total amount of phenolic compounds of fresh Gilaburu fruits and dried ones at different drying temperatures were determined and given in Table 4.

In convectional drying, an increase in the total phenolic content was observed with the shortening of the process time as a result of temperature elevation. It is thought that phenolic compounds of dried fruits were protected from thermal damage due to less exposure time to severe heating effect as a result of accelerated drying process with temperature increase.

Table 2. AE and rehydration rate values of dried Gilaburu fruits

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Drying	ΔE Values	Rehydration Rates		
60°C	$11.70 \pm 0.79^{a}$	$2.76 \pm 0.054^{\rm a}$		
70°C	$9.97 \pm 1.11^{a}$	$2.89 \pm 0.173^{\rm a}$		
80°C	$10.23 \pm 0.825^{a}$	$2.90 \pm 0.026^{\rm a}$		

Table 3. Textural properties, titration acidity (%), and pH values of fresh and dried Gilaburu fruits

Drying	Skin-hardness Values (g)	Elasticity Values (mm)	Titration Acidity (%) Values	pH Values
60 °C	$652 \pm 264^{a.b}$	$2.86 \pm 0.154^{b}$	$6.49 \pm 0.032^{b}$	$3.27 \pm 0.01^{b}$
70 °C	$687 \pm 195^{a.b}$	$2.78 \pm 0.126^{b}$	$6.89 \pm 0.168^{a}$	$3.25 \pm 0.005^{b.c}$
80 °C	$959 \pm 138^{a}$	$2.35 \pm 0.104^{c}$	$6.81 \pm 0.082^{a.b}$	$3.22\pm0.02^{c}$
Fresh Fruit	$325.3 \pm 27.4^{b}$	$4.38\pm0.139^{\mathrm{a}}$	$1.06 \pm 0.011^{c}$	$3.61 \pm 0.005^a$

Table 4. Total amount of phenolic compounds and antioxidant activity values of fresh and dried Gilaburu fruits

Drying	Total Phenolic Compounds (mg GAE g-1 DM)	Antioxidant Activity (mg TE g-1 DM)
60 °C	$35.31 \pm 2.45^{b}$	$72.77 \pm 0.775^{\circ}$
70 °C	$38.18 \pm 1.61^{a.b}$	$83.33 \pm 0.182^{b}$
80 °C	$40.95 \pm 2.12^{\mathrm{a.b}}$	$95.06 \pm 2.55^{a}$
Fresh Fruit	$45.46 \pm 2.86^{a}$	$55.25 \pm 2.64^{d}$

There are different studies in the literature that support our results. For example, in a study about cranberry drying, the loss of total phenolic content in samples dried at 50°C, 60°C and 70°C compared to fresh fruit was determined to be 51.25%, 47.13 and 38.56%, respectively. It has been reported that the reason for the total phenolic loss being higher for drying at 50 °C than that at 70°C is due to longer exposure time heat (Demir et al., 2019). It has been also reported in the literature that heat treatment can increase the digestibility of foods and allow active ingredients to be dissolved and released from foods more easily. Additionally, it has been stated that prolonged heat treatment may cause degradation or chemical reactions of active substances, thereby reducing their effective concentration (Wang et al., 2019).

# **Antioxidant Activity**

Antioxidant activity values of fresh Gilaburus and dried Gilaburus obtained at different drying temperatures were calculated on mg/g dry matter, in terms of trolox equivalent. All calculated values were given in Table 4. In convectional drying, it was observed that there was a decrease in antioxidant activity with an increase in the drying time of Gilaburu fruits. It is known that there is a strong relation between phenolic content of food material and its antioxidant activity. Therefore, it was thought that a decrease in antioxidant activity was detected in dried fruit samples as a result of phenolic loss depending on processing conditions. It can also be seen from Table 4, there is a strong relation between phenolic content and antioxidant activity for our results. Maillard reactions, progressing especially under high temperature treatments, cause higher antioxidant activity of food materials due to its products. In our study one part of an increase in antioxidant activity with temperature elevation may be associated with this Maillard reaction's products.

Lower antioxidant activity value for fresh Gilaburu fruit was obtained compared to all dried samples. This can be attributed to the easier dissolution of the active compounds in the extraction solvent and faster release from the food matrix under the influence of heat during the drying process. In addition, it has been reported that Maillard reaction's products produced especially during high temperature heat treatments are also effective in increasing antioxidant activity values (Tan et al., 2013). This does not applicable for fresh samples.

# Conclusion

Within the scope of our study, the effects of convectional drying method at different temperatures on the drying time and quality of Gilaburu fruit were determined. Convectional drying was carried out at temperatures of 60, 70 and 80 °C. It cannot be found any significant change in most of the studied physical properties except for skin-hardness and elasticity. Textural properties altered with drying, as skin hardness increased, elasticity decreased. On contrary, content of bioactive compounds and their functionality varied in wide extent with drying conditions. As a conclusion drying at moderate temperature levels resulted in dried Gilaburu fruits with better textural properties, although their functionality was lower than and other physical properties (color change,

rehydration ratio) were identical to dried samples at higher drying temperatures.

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